# Phosphorus Speciation in the Coastal Sediments of Khawr Ash Shaibah Al-Masdudah: Coastal Lagoon in the Eastern Red Sea, Kingdom of Saudi Arabia

### Radwan Al-Farawati; Mohamed El Sayed; Yaser Shaban; Amru El-Maradney; and Mohammed Orif

Department of Marine Chemistry, Faculty of Marine Science, King Abdulaziz University, P.O. Box 80207, Jeddah 21589, Kingdom of Saudi Arabia.

#### ID # (2738)

Received: 20/04/2013 In-revised: 17/09/2013 Correspondent Author: Radwan Al-Farawati E-mail: rfarawati@kau.edu.sa

#### **KEYWORDS**

Phosphorus, Marine, Coastal, Sediment, Red Sea, Speciation.

#### ABSTRACT

The total phosphorus concentrations and its speciation were studied in the surface sediment of Khawr ash Shaibah al Maftuhah. The distribution pattern of total phosphorus in the sediment showed east-west gradient indicating different depositional environment of the lagoon. The largest pool of the sedimentary phosphorus was apatitic phospgorus (57%) followed by refractory organic phosphorus (42%). The impact of loosely adsorbed and exchangeable phosphorus (0.9 % of total phosphorus) and Metal oxide Phosphorus (0.6 % of total phosphorus) on the budget of total phosphorus in the sediments was minor. The concentration of dissolved phosphorus in the upper water column is most probably buffered by apatitic phosphorus. The low concentration of metal oxide phosphorus is suggested to be influenced by the calcareous nature of the lagoon. Possible transformation between apatitic and refractory organic phosphorus is inferred from the correlation of both parameters. The correlation of refractory organic carbon with total phosphorus in the sediments suggests in situ production due to biological activities.

أشكال الفوسفور في الرواسب الساحلية لخور الشعبية المفتوحة: لاجون ساحلي على شرق البحر الأحمر بالمملكة العربية السعودية رضوان الفرواتي، محمد السيد، ياسر شعبان، عمرو المرادني، و محمد عريف قسم الكيمياء البحرية، كلية علوم البحار،جامعة الملك عبد العزيز ص ب 80207 ، جده 21589، المملكة العربية السعودية (الباحث المراسل: رضوان الفرواتي، بريد الكتروني: rfarawati@kau.edu.sa)

> رقم المسودة: (2738) تاريخ استلام المسودة:2013/04/20 تاريخ المسودة المُعَدَلة:2013/09/17 الباحث المُرَاسِل: رضوان الفرواتي بريد الكتروني: rfarawati@kau.edu.sa

#### الكلمات الدالة

الفوسفور ،ساحل البحر ، ر واسب، البحر الأحمر ، الأشكال، توقعات

### المستلخص

تمت دراسة التركيز الكلي للفوسفور وأشكاله في الرواسب السطحية لخور الشعيبة المفتوحة. أظهر نمط التوزيع للفوسفور الكلي في الرواسب التدرج من الشرق إلي الغرب مما يدل علي البيئة الترسيبية المختلفة في البحيرة. الفوسفور الأباتيتي هو المكون الأعظم من أشكال الفوسفور المترسب (57%) يليه الفوسفور العضوي المقاوم (42%). تأثير الفوسفور السهل والقابل للتبادل (0.9%) والفسفور علي هيئة أكسيد العنصر (0.6%) علي ميزانية الفوسفور الكلي في الرواسب كانت ضئيلة. تركيزات الفوسفور في طبقة المياه العلوية ربما يتم تنظيمها بواسطة الفسفور الأباتيتي. التركيز المنخفض من الفوسفور علي هيئة أكسيد العنصر ربما يتأثر بالخاصية الكلسية للبحيرة. هناك إمكانية للتحول المتبادل للفوسفور الغوسفور العضوي المقاوم والتي يستدل عليها من العلاقة بين العاملين. العلاقة بين الفسفور العضوي المقاوم والفسفور الكلي في الرواسب تقترح إنتاج موضعي كنتيجة للنشاط البيولوجي.

# Introduction

Nutrients in seawater have an important role in controlling the primary productivity in the oceans that in turn has potential impact on the atmospheric carbon dioxide (McElroy, 1983). Phosphorus is one of nutrients that is considered as bio-limiting element on geological timescales (Smith, 1984); (Codispoti, 1989). Rivers are the major source of phosphorus to the oceans whereas the ultimate sink is marine sediments. The balance between two processes are the main factor controlling the level of phosphorus in seawater (Froelich *et al.*, 1982); (Ruttenberg, 1993).

The Red Sea has oligotrophic nature, so the concentration of nutrients is significantly low with highest values observed in the Southern Red Sea (Morcos, 1970); (Edward, 1987). Most of studies of phosphorus in the Eastern coast of the Red Sea were concentrated on the dissolved fraction in seawater (El-Ravis, 1998), (El-Ravis and Moammar, 1998), (El-Sayed, 2002); (Turki et al., 2002); (Al-Farawati, 2010), & (El-Sayed, et al., 2011). Little has been done on the forms of phosphorus in the marine sediments in the Eastern Red Sea. The phosphorus in marine sediment constantly interacts with dissolved phosphate ions in the water through adsorption/desorption and coprecipitation/dissolution processes. In this study, total sedimentary phosphorus was fractionated into four distinct pools:

- (i) Loosely adsorbed and exchangeable-associated phosphate.
- (ii) Iron and aluminum-associated phosphate.
- (iii) Calcium-associated phosphate/ apatite.
- (iv) Refractory organic phosphorus.

# **Materials and Methods**

#### (1) Study Area

The Al-Shoaiba area consists of two lagoons: Northern and Southern lagoons, also known as Khawr Ash Shaibah Al-Masdudah and Khawr Ash Shaibah Al-Maftuhah. This study was restricted to study phosphorus speciation in the sediment of Khawr Ash Shaibah Al-Maftuhah that covers an area of figure 1.



**Figure 1:** Geographical Positions of Sampling Stations in Khawr Ash Shaibah Al-Maftuhah.

The lagoon is located approximately 100 km South of Jeddah, close to Shoaiba desalination plant. Mangrove stands are spread along the Eastern and Northern parts. Water exchange is limited to a narrow inlet in the southern lagoon. Additionally, the lagoon suffers from the dry and hot climate and there is no rivers flow. As a result high salinity values were recorded, ranging between ~ 42 to 50 (Rasul *et al.*, 2010). Detailed study on the sediment types, sediment composition and mineralogy, turbidity levels, water quality and bathymetry was recently studied (Rasul *e. al.*, 2010).

#### (2) Sampling and Methods of Analysis

The surface sediment was collected using the Van veen grab from each station and stored in polythene bags, (figure 2).



**Figure 2:** Surface Distribution of Total Phosphorus in the Sediments of Khawr Ash Shaibah Al-Maftuhah.

The sediment samples were dried in air, homogenized and powdered in an agate mortar. The extraction procedure used in this study is based upon the scheme developed by (Aspila et al., 1976), and (De Lange 1992) for marine sediment with minor alteration. One gram of dried sample was extracted in 25ml of filtered low-phosphate seawater (0.1µM phosphate) in a shaker at room temperature for 24h, which represented the loosely adsorbed and exchangeable phosphorus  $(P_{ex})$ . The residue from (P<sub>ax</sub>)was washed with 30ml of Milli-Q water twice to remove any residual seawater. The residue from  $(P_{av})$  extraction was extracted with 25ml of 0.5 N NaOH solutions to give fraction of iron and aluminium-associated phosphorus (P<sub>Fe/Al oxides</sub>). The residue from NaOH extraction was then extracted with 25ml of 0.5 N HCl, which represented apatite phosphorus calcium-associated and  $(P_{apatitic})$ . Total phosphorus  $(P_T)$  in sediment was determined by igniting 0.5g of sediment at 550°C in a muffle furnace and it was extracted with 0.5 N HCl after shaking for 16h. The residual refractory organic phosphorus  $(P_{org})$  is then determined as the difference between total phosphorus and the sum of  $(P_{ex})$ ,  $(P P_{Fe/Al})$  and  $(P_{apatite})$ . A centrifuged at 4000rpm was applied for the extracts for 10min. The supernatant in each extract was diluted with Milli-Q in 30 ml polypropylene vials to a final volume of 25 ml. The concentration of extracted phosphorus was analyzed using the molybdenum blue spectrophotometric method (Murphy and Riley, 1962).

#### **Results and Discussion**

The suitability of a phosphorus fractionation method for characterisation of phosphorus forms depends on the environment, the material to be studied, and on the type of material for which the method was originally designed. For instance, a high abundance of clay minerals, organic matter, and calcareous minerals may affect the ability of extraction procedures to distinguish between different phosphorus forms (Ruttenberg 1992). The sediments in the study area are typically poor in organic matter, but high in calcium carbonate (Rasul *et al.*, 2010). Thus, it was anticipated that calcium carbonate forms would be of major

importance. However, and as it can be seen below, the organic fraction is also shown to be a major fraction in the present study.

#### (1)Total Phosphorus (P<sub>T</sub>)

The concentration of total phosphorus in surface sediments of Khawr ash Shaibah Al-Maftuhah ranged between 2.53  $\mu$ Mg-1 (St.S13) and 14.14  $\mu$ Mg-1 (St. S8) with an average of 7.47  $\mu$ Mg-1 (table 1).

**Table 1:** Concentrations of phosphorus species  $(\mu M g^{-1})$  in the surface sediments of Khawr ash Shaibah al Maftuhah.

Station	P <sub>T</sub>	P <sub>ex</sub>	P <sub>Fe/Al oxides</sub>	P <sub>apatitic</sub>	P <sub>org</sub>
S01	6.90	0.07	0.03	4.29	2.51
S02	8.99	0.06	0.05	4.68	4.20
S03	8.21	0.03*	0.03	4.36	3.79
S04	7.40	0.06	0.07**	4.43	2.84
S05	8.34	0.08	0.04	4.45	3.77
S06	7.94	0.08	0.06	4.78	3.02
S07	8.19	0.06	0.04	4.88	3.21
S08	14.14**	0.05	0.05	3.38	10.66**
S09	9.50	0.06	0.05	3.14	6.25
S10	7.13	0.07	0.04	2.58	4.44
S11	5.29	0.05	0.05	2.81	2.38
S12	9.95	0.04	0.04	3.5	6.37
S13	2.53*	0.04	0.02*	0.72*	0.75
S14	5.90	0.05	0.02	2.85	2.98
S15	3.58	0.05	0.04	3.17	0.32*
S16	5.68	0.06	0.04	4.4	1.18
S17	5.60	0.07	0.05	3.97	1.51
S18	9.47	0.05	0.05	7.01**	2.36
S19	7.16	0.13**	0.05	4.71	2.27
Mean	7.47	0.06	0.04	3.95	3.28
	±2.55	$\pm 0.02$	±0.01	±1.16	±2.38

(\*\*High Values / \*Low Values)

These values are close to the values reported for Edko Lagoon in the Mediterranean (Khalil, 2007), Burullus Lagoon and shallow waters of Mediterranean (Khalil,*et al.*, 2007) and Great Barrier Reef continental shelf (Monbet *et al.*, 2007); but lower than those reported for Maryout Lagoon in the Mediterranean (Abdallah, 2011). Dramatic gradients in the concentration of total phosphorus between Eastern and Western part of the Northern lagoon are evident (see figure 2). The surface concentrations decreased toward the western part indicating different depositional environments of phosphorus.

# (2)The Loosely Adsorbed and Exchangeable Phosphorus $(P_{ax})$

The loosely adsorbed and exchangeable phosphorus is shown to be related to bioavailability of phosphorus (Zhou *et al.*, 2001). Beside phosphate concentration, temperature and salinity has significant impact on the release of loosely adsorbed and exchangeable phosphorus. In Khawr ash Shaibah al Maftuhah, the ( $P_{ex}$ ) concentration ranged between 0.03 and 0.13  $\mu$ Mg-1 and with average of 0.06  $\mu$ Mg-1 (see, table 1). The highest ( $P_{ex}$ ) concentration was observed in sediments collected from St.19 (0.13  $\mu$ M g<sup>-1</sup>) at the entrance of the lagoon. Apart of the highest value at the entrance of the lagoon, the distribution of ( $P_{ex}$ ) showed decreases westward (figure 3).



**Figure 3:** Surface Distribution of Loosely Adsorbed and Exchangeable Phosphorus in the Sediments of Khawr Ash Shaibah Al-Maftuhah.

The loosely adsorbed phosphorus represents about 0.9% of the sedimentary inorganic phosphorus (table. 2). Percentage of (Par) ranged between 1% and 25% was suggested to be due to high degree of oversaturation (Penn et al., 1995). It is also suggested that high concentrations of adsorbed phosphorus in the sediment is the main factor controlling algal blooms as the macrophytes effectively reduce the  $(P_{ex})$  content in the bottom sediment (Guang-Wei et al., 2006). The highest values observed at the entrance could be attributed to the absence of macrophytes and analogue marine plants due to high current speed. However, the values of sedimentary adsorbed phosphorus in Khawr ash Shaibah al Maftuhah indicate that this sedimentary phosphorus reservoir could be a minor contributor in the biogeochemical cycle of phosphorus in the study area.

Table 2: Percentage of Phosphorus Species (%) in the						
Surface Sediments of Khawr Ash Shaibah Al-Maftuhah.						

Station	P <sub>ex</sub>	P <sub>Fe/Al oxides</sub>	Papatitic	P <sub>org</sub>
S01	1.0	0.4	62	36
S02	0.7	0.6	52	47
S03	0.4*	0.4	53	46
S04	0.8	0.9	60	38
S05	1.0	0.5	53	45
S06	1.0	0.8	60	38
S07	0.7	0.5	60	39
S08	0.4	0.4	24*	75**
S09	0.6	0.5	33	66
S10	1.0	0.6	36	62
S11	0.9	0.9	53	45
S12	0.4	0.4	35	64
S13	1.6	0.8	68	30
S14	0.8	0.3*	48	51
S15	1.4	1.1**	89**	09*
S16	1.1	0.7	77	21
S17	1.3	0.9	71	27
S18	0.5	0.5	74	25
S19	1.8**	0.7	66	32
Mean	$0.9\pm0.4$	0.6±0.2	57±16	$42 \pm 17$

\*\*High Values / \*Low Values)

This low values are probably the results of the presence of high apatitic phosphorus contents in the sediment that buffer the concentration of the loosely adsorbed and exchangeable phosphorus (see below). This hypothesis is confirmed by plotting the concentrations of loosely adsorbed and exchangeable phosphorus and apatitic phosphorus that showed a positive correlation, (figure 4). However, although the correlation is positive, it is insufficient to explain the low concentration of the loosely adsorbed and exchangeable phosphorus. Other environmental parameters should be considered to solve the behaviour the loosely adsorbed and exchangeable phosphorus.



**Figure 4:** Correlation between of Loosely Adsorbed and Exchangeable Phosphorus and Apatitic Phosphorus in the Sediments of Khawr Ash Shaibah Al-Maftuhah.

(3) Metal Oxide Phosphorus ( $P_{Fe/Al \text{ oxides}}$ ) The concentration of phosphorus bound to iron and aluminium oxides (P<sub>Fe/Al oxides</sub>) is significantly low in Khawr ash Shaibah al Maftuhah. The fraction of phosphorus that is extracted by NaOH solution is used for estimation for short and long-term available phosphorus in the sediment particularly at the sediment-water interface in the presence of anoxic environment (Zhou et al., 2001);& (Wang, 2008).

The  $(P_{Fe/Al \text{ oxides}})$  concentration in the sediment of Khawr Ash Shaibah Al- Maftuhah ranged between 0.02 and 0.07  $\mu$ Mg-1 with an average of 0.04  $\mu$ Mg<sup>-1</sup> (see, table 1). The relative contribution of  $(P_{Fe/Al \text{ oxides}})$ to sedimentary inorganic P ranged between 0.3 and 1.1% with an average of 0.6% (see, table 2). The highest value of  $(P_{Fe/Al \text{ oxides}})$  was observed at St. 4 (.07  $\mu$ Mg<sup>-1</sup>) whereas the lowest value was found in the farthest north at stations St. 13 and 14. The general distribution showed relatively higher values in the Eastern region in comparison to the Western region that could be attributed to the higher contents of clay minerals delivered by dormant wadis (figure 5).



Figure 5: Surface Distribution Metal Oxide Phosphorus in the Sediments of Khawr Ash Shaibah Al-Maftuhah.

The distribution pattern of  $(P_{Fe/Aloxides})$  is in parallel to the distribution pattern of the loosely adsorbed and exchangeable phosphorus (see, figure 4).

This iron and aluminium phosphorus fraction contributed 6%-23% of total phosphorus in the sediment in Burullus Lagoon and shallow waters of Mediterranean (Khalil, et al., 2007), 43% in Maryout Lagoon in the Mediterranean (Abdallah, 2011) and 9%-13% in Edko Lagoon in the Mediterranean (Khalil, 2007).

The reported values of  $(P_{Fe/Al \text{ oxides}})$  from various basins have shown wide variability with higher values observed in eutrophic non-calcareous environments (Penn et al., 1995). Since the nature of Khawr Ash Shaibah Al-Maftuhah is predominant with calcareous deposits, the contribution of  $(P_{Fe/Al oxides})$  to the total phosphorus in sediment is anticipated to be minor as indicated from our data.

# (4) Apatitic Phosphorus (P<sub>apatitic</sub>)

Apatitic phosphorus was among the most stable P forms in the sediments in oxic and anoxic environments, and therefore not easily released to the upper water column (Salomons and Gerritse, 1981); (Jensen et al., 1995); (Gonsiorczyk et al., 1998). The  $(P_{apatitic})$  in the sediment of Khawr Ash Shaibah Al-Maftuhah ranged between 1.72 and  $7.01\mu$ Mg-1 with a mean of  $3.95\mu$ Mg-1 (see, table 1). On the other hand, the relative contribution of (P<sub>anatitic</sub>) to sedimentary inorganic phosphorus ranged between 24% and 89% with an average of 57% representing the largest phosphorus reservoir in the sediment of Khawr ash Shaibah al Maftuhah (figure 6), (see table 2).



Surface Distribution of Apatitic Figure 6: Phosphorus in the Sediments of Khawr Ash Shaibah Al-Maftuhah.

The apatitic phosphorus fraction contributed to 50%-73% of total phosphorus in the sediment in Burullus Lagoon and shallow waters of Mediterranean (Khalil, et al., 2007), 22% in Maryout Lagoon in the Mediterranean (Abdallah, 2011) and 49%-55% in Edko Lagoon in the Mediterranean (Khalil, 2007).

Possible of transformation between apatitic and refractory organic phosphorus through formation of some phosphorus-containing authigenic minerals has been reported (Ruttenberg, 1990); (Ruttenberg & Berner, 1993); & (Delaney, 1998). Although in this study, the authigenic apatite phosphorus was not distinguished from detrital apatite phosphorus. The correlation between the two major fractions (*i.e.* apatitic and refractory organic phosphorus) is shown in (figure 7).



**Figure 7:** Correlation between of Refractory Organic Phosphorus and Apatitic Phosphorus in the Sediments of Khawr Ash Shaibah Al- Maftuhah.

In case of existent of transformation between the two fraction, the relation should show a negative and strong correlation. Although the regression line has a negative direction as it should be, the correlation was very weak suggesting that the transformation through formation of some phosphorus-containing authigenic minerals could occurs but at limited levels.

# (5) Residual Refractory Organic Phosphorus (P<sub>ore</sub>)

Refractory organic phosphorus concentration in Khawr ash Shaibah al Maftuhah sediments ranges from 0.32 to 10.66  $\mu$ Mg-1 (see, table 1). It was accounted of 42% of total sedimentary phosphorus and was the second largest pool of sedimentary phosphorus (see, table 2). Low concentrations of refractory organic phosphorus were found in the southern region, and concentration increases systematically northward (Figure 8).



**Figure 8:** Surface Distribution of Refractory Organic Phosphorus in the Sediments of Khawr Ash Shaibah Al-Maftuhah.

The maximum concentrations  $(10.66\mu$ Mg<sup>-1</sup>) of refractory organic phosphorus were found in the North Eastern region. The maximum concentrations were found in Northern coastal stations (S8 and S9). This maximum is in agreement with the maximum organic carbon content based on the study by (Rasul, *et.al.*, 2010). However, the spatial distribution of refractory organic phosphorus is scattered and showed variability in the lagoon. The percentage of refractory organic phosphorus measured in the sediments of Khawr Ash Shaibah Al-Maftuhah is almost similar to that found by (Koch *et al.*, 2001) and it was attributed to the impact of mangrove on the sediments.

The correlation between the concentration of refractory organic phosphorus and calcium carbonate content for sediments from Florida Bay was negative and strong indicating in situ biological production by sea grass and algae. The percentage of calcium carbonate was not determined in the present study. However, the distribution pattern of the refractory organic phosphorus in Khawr Ash Shaibah Al- Maftuhah, (see, figure 8), is almost inverse the distribution pattern of the percentage of calcium carbonate in Khawr Ash Shaibah Al- Maftuhah by the study of (Rasul, et al., 2010), indicating in situ biological production of refractory organic phosphorus. On the other hand, the correlation of total phosphorus and refractory organic phosphorus is positive and

strong indicating that a significant proportion of sedimentary phosphorus is produced by in situ biological activities (figure 9).



**Figure 9:** Correlation between of Refractory Organic Phosphorus and Total Phosphorus in the Sediments of Khawr Ash Shaibah Al-Maftuhah.

# Conclusion

The total phosphorus concentrations and its speciation were studied in the surface sediment of Khawr Ash Shaibah Al-Maftuhah Lagoon. Highest concentrations of total phosphorus were evident in the Eastern part. The majority of phosphorus was concentrated in two forms; apatitic phospgorus (57%) and refractory organic phosphorus (42%). The most important form that possibly controls the concentrations of dissolved phosphorus is apatitic phosphorus. The high correlations between apatitic and refractory organic phosphorus suggested transformation process of apatitic phosphorus to organic phosphorus and vice versa.

# Acknowledgement

This project was funded by the Deanship of Scientific Research (DSR), King Abdulaziz University, Jeddah, Kingdom of Saudi Arabia, under grant No.(301/150/431). The authors, therefore, acknowledge with thanks (DSR) technical and financial support. Map of the area was prepared by Dr. Ramadan Abo Zaid.

# References

Abdallah AMA (2011) Potential for Internal Loading by Phosphorus Based on Sequential Extraction of Surficial Sediment in a Shallow Egyptian Lake. *Environmental Monitoring and Assessment*, **178** (1-4): 203-212. Available at: http://www.link.springer.com/art icle/10.1007%2Fs10661-010-1682-2

- Al-Farawati R (2010) Environmental Conditions of the Coastal Waters of Southern Corinche, Jeddah, Eastern Red Sea Physico-Chemical Approach. *Australian Journal of Basic and Applied Sciences*, 4 (8): 3324-3337 Available at: https://www.kau.edu.sa/GetFile. aspx?id=108080&fn...%20Jeddah...pdf
- Aspila KI; Agemian H; and Chau ASY (1976) A Semi-Automated Method for the Determination of Inorganic: Organic and Total Phosphate in Sediments. *Analyst* 101: 187-197. Available at: http://www.pubs.rsc.org/e\_n/Co

Available at: http://www.pubs.rsc.org/e n/Co ntent/ArticleLanding/1976/AN/an976010018 7#!divAbstract

- Codispoti LA (1989) Phosphorus Vs. Nitrogen Limitation of New and Export Production. *In:* Berger WH; Smetacek, VS; and Wefer G (eds) *Productivity of the Ocean: Present and Past*: Report on the Dahlem Workshop on Productivity of the Oceans, Berlin 1988. John Wiley and Sons, Berlin, Germany, pp377-394. Available at: http://www.onlinelibrary.wiley. com/doi/10.1002/gj.3350260119/abstract
- Delaney ML (1998) Phosphorus Accumulation in Marine Sediments and the Oceanic Phosphorus Cycle. *Global Biogeochem. Global Biogeochemical Cycles*, **12** (4): 563-572. Available at: http://www.onlinelibrary.wiley. com/doi/10.1029/98GB02263/abstract
- **De Lange GD** (1992) Distribution of Various Extracted Phosphorus Compounds in the Interbedded Turbiditic/Pelagic Sediments of the Maderia Abyssal Plain, Eastern North Atlantic. *Marine Geology*, **109** (1/2): 115-139. Available at: http://www.sciencedirect.com/ science/article/pii/0025322792902246
- **Dos Santos Afonso M;** and **Stumm W** (1992) Reductive Dissolution of Iron (III) (hydr) Oxides by Hydrogen Sulfide. *Langmuir*, **8** (6): 1671-1676.

Available at: http://www.pubs.acs.org/doi/abs/10.1021/la00042a030

- Edwards AJ (1987) Climat and Oceanography. In: Edwards AJ; and Head SM (eds.) Key Environments: Red Sea. Pergamon Press, Oxford, UK, pp 45-69.
- El-Rayis OA; and Moammar MO (1998) Environmental Conditions of Two Red Sea Coastal lagoons in Jeddah: 1. Hydrochemistry. *Journal of King Abdulaziz University: Marine Sciences*, 9 (1): 33-45.

Available at: http://www.journaldatabase.org/ articles/environmental\_conditions\_two\_red\_ sea.html

- El-Rayis OA (1998) Environmental Conditions of Two Red Sea Coastal Lagoons in Jeddah: 2- Nutrients. *Journal of King Abdulaziz University : Marine Sciences*, 9 (1): 49-59. Available at: http://www.journaldatabase.org/ articles/environmental\_conditions\_two\_red\_ sea\_coastal.html
- El-Sayed MA (2002) Distribution and Behavior of the Dissolved Species of Nitrogen and Phosphorus in Two Coastal Red Sea Lagoons Receiving Domestic Sewage. *Journal of King Abdulaziz University: Marine Sciences*, **13** (1): 47-73.

Available at: http://www.journaldatabase.org/ articles/distribution behavior dissolved.html

- El-Sayed MA; El-Maradny AA; Al-Farawati R; and Shaban YA (2011) Evaluation of the Adequacy of a Rehabilitation Programme, Implemented in Two Red Sea Coastal Lagoons, Using the Hydrological Characteristics of Surface Water. *Journal of King Abdulaziz University: Marine Sciences*, 22 (2): 69-108. Available at: http//www.prod.kau.sa/centers/spc/ jkau/Data2/Review\_Article.aspx?No=33 05
- Froelich PN; Bender ML; Luedtke NA; Heath, GR; and De Vries T (1982) The Marine Phosphorus Cycle. American Journal of Science, 282 (4): 474-511. Available at: http://www.ajsonline.org/conte n

t/282/4/474.full.pdf+html

Gonsiorczyk T; Casper P; and Koschel R (1998) Phosphorus Binding Forms in the Sediment of an Oligotrophic and an Eutrophic Hardwater Lake of the Baltic District (Germany). *Water Sciene and Technol*ogy, **37** (3): 51-59. Available at: http://www.iwaponline.com/wst/03703/wst037030051.htm

Guang-Wei Z; Bo-Qiang Q; Lu Z; and Lian-Cong L (2006) Geochemical forms of Phosphorus in Sediments of Three Large, Shallow Lakes of China. *Pedoqhere*, **16** (6): 726-734.

Available at: http://www.sciencedirect.com/sc ience/article/pii/S1002016006601082

Khalil MKH; Radwan AM; and El-Moselhy KHM (2007) Distribution of Phosphorus Fractions and Some Heavy Metals in Surface Sediments in Burullus Lagoon and Adjacent Mediterranean Sea. *Egyptian Journal Aquatic Research*, 33 (1): 277-289.

Available at: http://www.oceandocs.org/ bitstream/1834/1888/1/Text.pdf

Khalil MKH (2007) The Fractional Composition of Phosphorus in Edku Lagoon and Adjacent Marine Sediments, Egypt. Journal of King Abdulaziz University: Marine Sciences, 19 (1): 149-166.

Available at: http://ecc.isc.gov.ir/ShwFArticle. aspx?aid=130461

Koch MS; Benz RE; and Rudnick DT (2001) Solid-phase Phosphorus Pools in Highly Organic Carbonate Sediments of Northeastern Florida Bay. *Estuarine Coastal and Shelf Science* 52 (2): 279-291. Available at: http://www.adsabs.harvard.edu/

Available at: http://www.adsabs.harvard.edu/ abs/2001ECSS...52..279K

McElroy MB (1983) Marine Biological Controls on Atmospheric CO<sub>2</sub> and Climate. *Nature* **302** (5906): 328-329. Available at: http://www.nature.com/nature/

journal/v302/n5906/abs/302328a0.html

- Monbet P; Brunskill GJ; Zagorskis I; and Pfitzner J (2007) Phosphorus Speciation in the Sediment and Mass Balance for the Central Region of the Great Barrier Reef Continental Shelf (Australia). *Geochimica et Cosmochimica Acta* 71 (11): 2762-2779. Available at: http://www.sciencedirect.com/ science/article/pii/S0016703707001640
- Murphy J; and Riley JP (1962) A Modified Single Solution Method for the Determination of Phosphate in Natural Waters. *Analytica*

Chimica Acta 27 (1): 31-36.

Available at: http://www.sciencedirect.com/ science/article/pii/S0003267000884445

Penn MR; Auer MT; Van Orman EL; and Korienek JJ (1995) Phosphorus Diagenesis in Lake Sediments: Investigation using Fractionation Techniques. *Marine and Freshwater Res*earch, **46** (1): 89–99.

Available at: http://www.publish.csiro.au/ paper/MF9950089.htm

Rasul NMA; Al-Hazmi OA; Abushosha MI; Al-Nomani SM; Akbar RR; Bahareth FZ; Widinly NA; Al-Farawati RK; Bantan RA; and Qutub AS (2010) Sedimentological and Geochemical Properties and the Application of Satellite Images in the Two Shaibah Lagoons and the Associated Sabkhas, Red Sea Coast of Saudi Arabia. *Saudi Geological Survey, Technical Report (SGS-TR-2008-9),* pp114. Available at: http://www.kau.edu.sa/Content. aspx?Site ID=0002674&lng=EN&cid=17435

Ruttenberg KC; and Berner RA (1993) Authigenic Apatite Formation and Burial in Sediments From Non-Upwelling, Continental Margin Environments. *Geochimica et Cosmochimica Acta* 57 (5): 991–1007. Available at: http://www.sciencedirect.com/ science/article/pii/001670379390035U

Ruttenberg KC (1992) Development of a Sequential Extraction Method for Different Forms of Phosphorus in Marine Sediments. *Limnology and Oceanography*, **37** (7): 1460-1482.

Available at: http://www.research.mblwhoil ibrary.org/works/18920

Ruttenberg KC (1990) Diagenesis and Burial of Phosphorus in Marine Sediments: Implications for the Marine Phosphorus Budget. Faculty of the Graduate School, Yale University in Candidacy, New York, USA, PP 1-750. (PhD Dissertation).

Available at: http://www.books.google.com. bh/books/about/Diagenesis\_and\_burial\_of\_ phosphorus\_

Ruttenberg KC (1993) Reassessment of the Oceanic Residence Time of Phosphorus. *Chemical Geology*,**107** (3/4): 405-409.

Available at: http://www.sciencedirect.com/ science/article/pii/000925419390220D

- Salomons W; and Gerritse RG (1981) Some Observations on the Occurrence of Phosphorus in Recent Sediments from Western Europe. *Science of the Total Environment*, **17** (1): 37-49. Available at: http://www.edepot.wur.nl/212245 Available at: http://www.sciencedirect.com/ science/article/pii/0048969781901066
- Smith SV (1984) Phosphorus Versus Nitrogen Limitation in the Marine Environment. *Limnology and Oceanography*, **29** (6): 1149-1160.

Available at: http://t.tube.aslo.net/lo/toc/vol\_2 9/issue\_6/1149.pdf

Zhou Q; Gibson CE; and Zhu Y (2001) Evaluation of Phosphorus Bioavailability in Sediments of Three Contrasting Lakes in China and the UK. *Chemosphere*, 42 (2): 221-225.
Available at: http://www.sciencedirect.com / science/article/pii/S0045653500001296